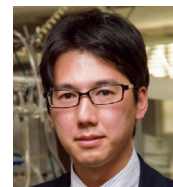




Development of high-performance CO₂ electrolysis via rational design of triple-phase interfaces

Research Center for Solar Energy Chemistry, Graduate School of Engineering Science

Associate Professor Kazuhide Kamiya

<https://researchmap.jp/kamiya0908?lang=en>

Abstract

The electrochemical reduction of CO₂ proceeds under ambient conditions and is therefore attracting considerable attention as a technology for CO₂ valorization. To achieve high-current-density electrolysis of gaseous CO₂, it is essential to elucidate the reaction mechanisms occurring at the triple-phase interface composed of the catalyst, electrolyte, and CO₂ gas, and to establish rational design principles. We conducted a multiscale and multimodal investigation focusing on the triple-phase interface and identified the interparticle spacing within the catalyst layer as a key parameter governing the achievable current density. Furthermore, we demonstrated that the size of cations in the electrolyte critically determines product selectivity through modulation of the electric double-layer environment.

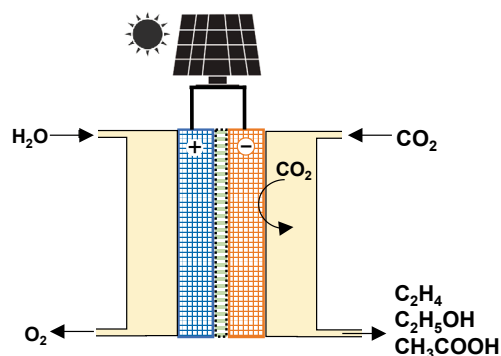
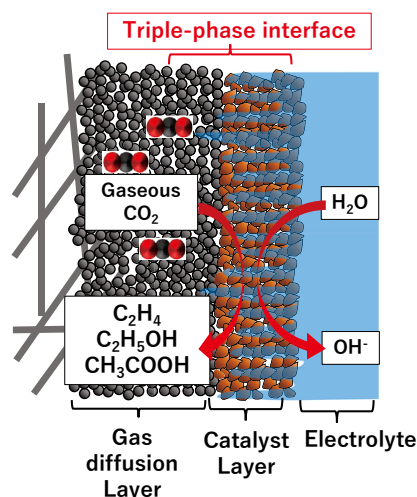
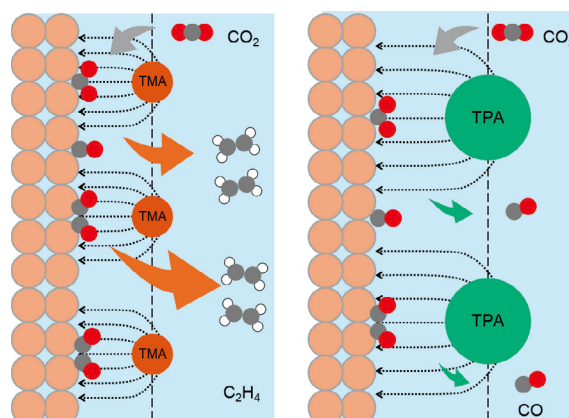
Background & Results

The development of technologies that reduce and convert CO₂ is essential to close the carbon cycle. The electrochemical CO₂ reduction reaction has gained prominence because it operates under ambient conditions. In traditional CO₂ electrolysis, an immersed electrode setup has generally been used. In such systems, the current density is limited by the solubility and mass transport of CO₂ in the solution. Gas diffusion electrodes have been employed to overcome these transport limitations. Supplying gaseous CO₂ directly to the catalyst layer enables high current density electrolysis. Nevertheless, gaseous CO₂ reduction proceeds at a complex triple-phase interface formed by the catalyst, electrolyte, and gaseous CO₂. The fundamental reaction mechanisms occur at this interface and the corresponding design principles for achieving high performance remain insufficiently elucidated.

We have pursued the activation of the triple-phase interface by analyzing the underlying physicochemical phenomena using a multiscale and multimodal approach that combines experiments and simulations. First, we elucidate the fundamental requirements for the high-current-density CO₂ reduction reaction by systematically comparing various physicochemical properties of electrodes and their corresponding CO₂RR activities. We identified that the average size of interparticle spacing in the catalyst layer is correlated with the maximum partial current density for C₂₊ production. Furthermore, we demonstrated that the hydrated radius of the electrolyte cations strongly influences the selectivity toward C₂₊ products. In particular, smaller cations generate stronger interfacial electric fields within the electric double layer, which promote more efficient formation of multi-carbon products such as ethylene and ethanol.

Significance of the research and Future perspective

This study targets the mesoscopic reaction environment that bridges these scales, namely the design of the triple-phase interface. Interface engineering enabled the simultaneous attainment of state-of-the-art current densities and high selectivity toward C₂₊ products. Future efforts will integrate materials design and systems engineering with interfacial design to further enhance the performance of gaseous CO₂ electrolysis.

CO₂ electrolysis in a membrane-electrode-assembly electrolyzerTriple-phase interface for gaseous CO₂ electrolysis

Electric field variation in the electric double layer by cation size

Patent Japanese Patent No.7641381, Japanese Unexamined Patent Publication No.2024-113569**Treatise** Kurihara, Ryo; Nakanishi, Shuji; Kamiya, Kazuhide et al. Alkali-cation-free electrochemical CO₂ reduction to multicarbon products in aqueous electrolytes containing tetraalkylammonium cations. *EES Catalysis*. 2025, 3, 1055–1061. doi: 10.1039/D5EY00141B
Inoue, Asato; Nakanishi, Shuji; Kamiya, Kazuhide et al. Small interparticle spacing in catalyst layers forms an expansive triple-phase interface for boosting the current density of CO₂-to-C₂₊ conversion. *Small*. 2025, 21(23), e2500693. doi: 10.1002/smll.202500693**U R L****Keyword** CO₂ electrolysis, net zero emissions, triple-phase interface, membrane electrode assembly, artificial photosynthesis